

BSc. (HONS) - PHYSICS ADMISSION TEST

INSTRUCTIONS

- The Entrance Test will be of 60 mins
- The booklet has three Sections

PHYSICS	15 Questions (15 X 1 = 15)
MATHEMATICS	15 Questions (15 X 1 = 15)
CHEMISTRY	15 Questions (15 X 1 = 15)
ENGLISH	05 Questions (05 X 1 = 05)

- All questions carry equal marks.
- For each correct answer 2 marks will be awarded. There is no negative marking.
- Answer to be given by ticking the correct option.
- Use ball pen or dot pen only.
- All the rough work should be done on the space provided in the booklet.

MM- 50 Time- 60 Mins

Details of the candidate

Name
Father's Name
Course Applied for

PHYSICS

1. The particles which can be added to the nucleus of an atom without changing its chemical properties are
A. Electrons
B. Protons
C. Neutrons
D. Positron
2. Who gave the Quantum model of hydrogen atom?
A. S.N Bose
R Neils Rohr

3. What is the unit of Astronomical Distance?

C James Clerk Maxwell

A. Light Year

DR.A Millikan

- B. Angstrom
- C. Weber
- D. Lux
- 4. If no external force acts on a system of bodies, the total linear momentum of the system of bodies remains constant. Which law states that ?
 - A. Newton's first law.
 - B. Newton's Second Law
 - C. Newton's Third Law.
 - D. Principle of conservation of linear momentum
- 5. An air bubble in water will act like a
 - A. convex lens
 - B. convex mirror
 - C. concave lens
 - D. concave mirror
- 6. with the increase of pressure, the boiling point of any substance
 - A. Increases
 - B. Decreases
 - C. Remains Same
 - D. Becomes zero
- 7. The phenomenon of interference is based on
 - A. Conservation of momentum.
 - B. conservation of energy.
 - C. conservation of momentum and energy.
 - D. quantum nature of light.
- 8. The electric motor converts
 - A. Electrical energy into mechanical energy
 - B. Mechanical energy into electrical energy
 - C. Electrical energy into light energy
 - D. None of these

 9. If a lift is going up with acceleration, the apparent weight of a body is A. More or less the true weight B. Equal to the true weight C. Less than the true weight D. More than the true weight 				
10. If 6	electrical conductivity in A. Conductor	ncreases with the increase of ter B. Semiconductor	nperature of a substanc C. Insulator	e, then it is a: D. Carbonator
11. W	•	tion of Light.	ly to the reddish appear	ance of the sun at sunrise or sun
	wo bulbs of power 25 W Which bulb will fuse? A. 25 W Bulb B. 40 Watt C. None of these. D. Both 1 & 2.	and 100 W respectively each ra	ited at 220 V are connec	ted in series with the supply of
		re carries downward electric cu narges moving downward instea		
require	d to stop the ejection o	uminated by a monochromatic I f electrons is 3v. When the same equired to stop the ejection of e	e surface is illuminated l	by the light of wavelength 2λ,
	•	action of copper is 4.47 eV.The n hanged when irradiated by light	•	
	A. 4.37V B. 4.23V C. 8.70V D. 8.46V			

MATHEMATICS

1. Solve the inequality: -(x+2)+2x> A No Solution	2(x-3)+3x B (4/5,+∞)	C. (-∞+∞)	D. (-∞, 1)
2. Which of the following do(es) not	belong to A×B for the sets A={1.2	?} and B={0.2}?	(, ,
A R={(1, 1),(2, 1)} C. R={(1,0),(1,2)} 3. If A is a square matrix such that A	B. R={(1,0),(2,2)} D. R={(1,2),(2,2)}	, (4,=, .	
A. 3I	В. О	C. 21	D. I
4. Which of the following is not irratA. (2 – √3)2	ional? B. (V2 + V3)2	C. (V2 + V3)	D. 27√7
71. (Z V3)Z	D. (VZ 1 V3)Z	C. (V2 · V3)	D. 27 V7
5. If the line x+2ay+a=0,x+3by+b=0,x+4cy+c=0 are concurrent, then a,b,c are in. A Arithmetic Progression B Geometric Progression C Harmonic Progression D Arithmetic Geometric Progression			
6. LCM of 2 ³ × 3 ² and 2 ² × 3 ³ is A 2 ³	B. 2 ³ × 3 ³	C. 3 ³	D. $2^2 \times 3^2$
A Z	B. 2° × 3°	C. 3°	D. 2" × 3"
7. The value of the determinant $\begin{bmatrix} 1 \\ 2 \\ 3 \end{bmatrix}$	0 0 4 0 is 5 2		
A. 0	B. 8	C. 7	D. 5
8. If z= 2-4i then complex conjuga A. 2+4i	te of z is B2+4i	C2-4i	D. 2-4i
9. What will be the length of the s	ide BC in a right angle triangle Al	BC if AB= 12 cm and $\Delta A = 30^{0}$ 8	$\Delta C = 90^{\circ}$
A. 24	B. 6√3	C. 6	D 24√3
10. The mode of the following data : A 14	14, 20, 27, 20, 14, 14, 13, 13, 21, B. 20	10, 27 is C. 13	D. 27
.11. If sinΘ = x and secΘ = y , then tag A. xy	n Ө is В. x/y	C. y/x	D. 1/xy
12. The projection of a line segment segment is	on the axes of reference are 3,4	and 12 respectively. The length o	of the line
A 19	B. 19/3	C. 5	D. 13
13. The cost of cultivating a square fit around it at the rate of 75paise per n	neter would be	·	
A Rs 360	B. Rs 810	C. Rs 900	D. Rs 1800
14. The differential equation of all st A y-1=mx	raight lines in a plane passing thi B. y=m(x-1)	rough (0, 1) is: C. y= xy ₁	D. y=x ₁ +1
15.The D.E whose solution is y = Asin	2x + B cos 2x given as:		
A y ₂ =4y	B. $y_2+4y=0$	C .y ₂ +3 y=0	D. $y_2+y=0$

CHEMISTRY

1.	Which of the following is no	2	0.01	
	A. Potassium Chloride	B. Cesium chloride	C. Glass	D. Rhombic Sulphur
2.	An example of fossil fuel is A. Coal	B. Animal waste	C. Wood	D. All of these
3.	An electrolyte A a metal	B. a solution	C. a liquid that conducts currer	nt D. all of above
4.	The process of transferring cha A. Transferring	B. Processing	arth is called: C. Charging	D. Earthling
5. I	How many chloride ions are su A 4	rrounding sodium ion in sodi B8	um chloride crystal ? C 6	D 12
6. I	Bakelite is an example of A elastomer	B fibre	C thermoplastic	D thermosetting
7.	Green House gases includes: A. CO ₂ , Methane, Nitrous ox C. CO ₂ , Methane, Chorine	ide B. CO ₂ , Argon, D. CO ₂ , Metha		
8. /	A substance added to food con A. Oxidant	taining fats and oils is called B. Rancid	: C. Coolant	D. Antioxidant
9.	The drugs which are given to th	e patients suffering from an	xiety and mental tension are kn	own as
	A. tranquilizers	B. Analgesics	C. Antimicrobials	D antibiotics
10	. Heat conduction is the prope	rty of		
	A. Non-metal	B. Metal	C. Metalloids	D. All of these
	 Oxidation is a process which i A. addition of oxygen 	nvolves B. Addition of hydrogen	C. removal of oxygen	D. removal of hydroger
	. Pure water can be obtained f A. Centrifugation	rom sea water by: B Plasmolysis	C Reverse osmosis	D Sedimentation
13.	. Solubility of a gas in a liquid in	ncreases on:		
	A. Increasing temperature.B. Decreasing pressure.C. Increasing pressure.D. Increasing temperature and	pressure.		
E	A. Which one is the most electric A. Sugar dissolved in water B. Salt water C. Salt dissolved in an organic so D. An oil and water mixture		lowing choices?	
A	The boiling point of alcohol is A. hydrogen bonding B. large size of alcohol	higher than ether due to:		

C. presence of -OH group D. high molecular weight

ENGLISH

FILL IN THE BLANKS

Choose the most appropriate word to fill the blank

1. The group was found to	be in subversive activities.
A. involved	
B. Engaged	
C. Rebellious	
D. uncharitable	
	en on any evidence he gave.
A. challenged	
B. cross- examined	
C. Praised	
D. questioned	
3. The ship,	Robinson arrived on the Island.
A had been broken	
B Having been brok	en
C. having broken D. Has broken	
D. Has blokell	
4. John failed his exams be	ecause he was always out with his friends when he
A should have beer	n studying
B. used to study	
C. must have been	: =
D. will have studied	
	mal to leave the room and him to return.
A. Stopped	
B. Refused C. Forbade	
D. Challenged	
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BSc (Hons) MATHS ADMISSION TEST

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Details of the candidate

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PHYSICS

1. Nuclear sizes are expressed in a unit named	
(a) Fermi	(b) Angstrom
(c) Volt	(d) Tesla
2. The speed of light will be minimum while passing through	
(a) Water	(b) vacuum
(c) Air	(d) Glass
3. Which of the following is not a vector quantity?	
(a) Speed	(b) velocity
(c) Torque	(d) displacement
(c) rorque	(u) displacement
4. A pencil placed vertically on a table falls down. What will be	the linear velocity of middle of pencil
at the end of the fall if the pencil is 15 cm long?	
(a) 1.05 m/s	(b) 2.1 m/s
(c) 3 m/s	(d) 0.5 m/s
(6) 5, 5	(4) 0.5 11,5
Finally to the falls that the control of the base of the task	
5. Which of the following laws was modified by Maxwell by int	• .
(a) Gauss's law	(b) Ampere's law
(c) Biot-Savart's law	(d) none of these
6. When light travels from one medium to another, which of the	ne following does not change?
(a) Frequency	(b) velocity
(c) refractive index	(d) wavelength
7 M/high of the following is consequed when light waves intent	aua?
7. Which of the following is conserved when light waves interf	
(a) Phase	(b) intensity
(c) Amplitude	(d) none of these
8. When a Polaroid is rotated, the intensity of light varies but r	never reduces to zero. It shows that
the incident light is	
(a) Unpolarised	(b) completely plane polarised
(c) partially plane polarised	(d) None of the these
	. ,
9. The principle that a quantum orbital cannot be occupied by	more than two electrons was given
by:	
(a) Pauli	(b) Millikan
(c) Hund	(d) None of these
40 M/hat is massaum duritals the North court of	
10. What is measured with the Nephometer?	
(a) Volume of rainfall	(b) Cloud volume and speed
(c) Salinity of Sea	(d) All options are correct.

CHEMISTRY

1. Which one of the following is non-crystalline or amor	phous?
(a) Diamond	(b) Graphite
(c) Glass	(d) Common Salt
2. The atmospheric pollution is generally measured in th	ne units of
(a) Mass percentage	(b) volume fraction
(c) Volume percentage	(d) ppm
3. Hydrogen bomb is based on the principle of	
(a) Nuclear fission	(b) nuclear fusion
(c) Natural radioactivity	(d) artificial radioactivity
4. Faraday's law of electrolysis is related to	
(a) Atomic number of cation	(b) Speed of cation
(c) Speed of anion	(d) Equivalent weight of electrolyte
 In a reaction, 2X → Y, the concentration of X decrease 	es from 0.50 M to 0.38 M in 10 min. What is
the rate of reaction in Ms ⁻¹ during this interval?	
(a) 2×10^{-4}	(b) 2×10^{-2}
(c) 4×10^{-2}	(d) 1×10^{-2}
6. Which of the following types of metals make the mos	t efficient catalyst?
(a) Alkali metals	(b) Transition metals
(c) Inner transition metals	(d) Alkaline earth metals
7. Which of the following is a non-metal that remains liq	quid at room temperature?
(a) Chlorine	(b) Phosphorous
(c) Bromine	(d) Helium
8. Which of the following amides will give ethylamine or	n reaction with sodium hypobromide?
(a) Butanamide	(b) Acetamide
(c) Propanamide	(d) Benzamide
9. Which one of the following is not a mixture?	
(a) Air	(b) Mercury
(c) Milk	(d) Cement
10. What is the basic formulae for starch?	
(a) $(C_6H_{12}O_6)_n$	(b) $C_{12}O_{12}O_{11}$
(c) $(C_6H_{10}O_5)_0$	(d) $(C_6H_{12}O_4)_n$

MATHEMATICS

1. If an operation is defined by $a^* b = a^2 + b^2$, then $(1 * 2) * 6$ is	
(a) 12	(b) 28
(c) 61	(d) None of these
2. Simplified form of cos ⁻¹ (4x ³ – 3x)	
(a) 3 sin ⁻¹ x	(b) 3 cos ⁻¹ x
(c) $\pi - 3 \sin^{-1}x$	(d) None of these
3. If A and B are square matrices then (AB)' =	
(a) B'A'	(b) A'B'
(c) AB'	(d) A'B'
4. The area of a triangle with vertices (-3, 0) (3, 0) and (0, k) is 9	sq. units. The value of k will be
(a) 9	(b) 3
(c) -9	(d) 6
5. What type of a relation is R = {(1, 3), (4, 2), (2, 4), (2, 3), (3, 1))} on the set A – {1, 2, 3, 4}
(a) Reflexive	(b) Transitive
(c) Symmetric	(d) None of these
6. The set of points where the function f given by $f(x) = 2x - 1 $	sin x is differentiable is
(a) R	(b) (0, ∞)
(c) $R = \{1/2\}$	(d) None of these
7. $(2, -3, -1)$ 2x - 3y + 6z + 7 = 0	
(a) 4	(b) 2
(c) 3	(d) 1/5
8 Area of the region in the first quadrant enclosed by the x-axis = 32 is	s, the line $y = x$ and the circle $x^2 + y^2$
(a) 16π sq. Units	(b) 4π sq. units
(c) 32π sq. Units	(d) 24π sq. units
9. The maximum value of the object function Z = 5x + 10 y subj	ect to the constraints $x + 2y \le 120$, x
$+ y \ge 60, x - 2y \ge 0, x \ge 0, y \ge 0$ is	
(a) 300	(b) 600
(c) 400	(d) 800
10. If A, B are two events associated with same random experi and $P(B/A) = 0.6$ then $P(A/B)$ is	ment such that P(a) = 0.4, P(b) = 0.8
(a) 0.3	(b) 0.5
(c) 0.4	(d) 0.6

ENGLISH

FILL IN THE BLANKS

1. Our Sir teaches Mathematics English. (a) Across (c) Beside	(b) Besides (d) Both
2. I don't know the city he lives. (a) What (c) Where	(b) when (d) which
3. After six months, you can also speak in English(a) Around	me. (b) like
(c) Without 4. America a powerful president (a) Have	(d) about (b) is
ONE WORD SUBSTITUTION	(d) does
5. A story of old times gods or heroes:(a) Lyric(c) Epic	(b) Legend (d) Romance